

- Johnson, C. K. (1976). ORTEPII. Report ORNL-5138. Oak Ridge National Laboratory, Tennessee, USA.
- North, A. C. T., Phillips, D. C. & Mathews, F. S. (1968). *Acta Cryst. A* **24**, 351–359.
- Preetz, W., Peters, G. & Bublitz, D. (1996). *Chem. Rev.* **96**, 977–1025.
- Preetz, W. & Ruf, D. (1986). *Z. Naturforsch. Teil A*, **41**, 871–878.
- Preetz, W., Ruf, D. & Tensfeldt, D. (1984). *Z. Naturforsch. Teil B*, **39**, 1100–1109.
- Sheldrick, G. M. (1990). *Acta Cryst. A* **46**, 467–473.
- Sheldrick, G. M. (1993). *SHELXL93. Program for the Refinement of Crystal Structures*. University of Göttingen, Germany.

*Acta Cryst.* (1997). **C53**, 66–67

### [1,2-Bis(diphenylphosphino)ethane-*P,P'*]-bis( $\alpha$ -toluenethiolato-S)palladium(II), [Pd{Ph<sub>2</sub>P(CH<sub>2</sub>)<sub>2</sub>PPh<sub>2</sub>}](SCH<sub>2</sub>Ph)<sub>2</sub>]

WEIPING SU, MAOCHUN HONG,\* RONG CAO AND  
HANQIN LIU

*State Key Laboratory of Structural Chemistry, Fujian Institute of Research on the Structure of Matter, Fuzhou, Fujian 350002, People's Republic of China*

(Received 16 April 1996; accepted 27 September 1996)

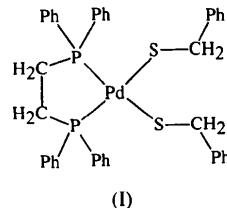
#### Abstract

The title compound, [Pd(C<sub>7</sub>H<sub>7</sub>S)<sub>2</sub>(C<sub>26</sub>H<sub>24</sub>P<sub>2</sub>)], is a mononuclear palladium(II) complex. The molecule possesses a crystallographic twofold axis and the Pd atom is four-coordinated by two phosphine P atoms and two S atoms from PhCH<sub>2</sub>S<sup>−</sup> ligands, and has a distorted square-planar geometry. The Pd—S and Pd—P distances are 2.360 (2) and 2.277 (2) Å, respectively.

#### Comment

Transition metal compounds with mixed sulfur and phosphine ligands have attracted much attention due to their relevance and importance to a wide variety of chemical and industrial systems. Of the nickel group metals, many nickel compounds with such mixed ligands have been reported. Surprisingly few palladium compounds, such as [Pd<sub>2</sub>(SC<sub>6</sub>F<sub>5</sub>)<sub>2</sub>(PPh<sub>3</sub>)<sub>2</sub>] (Fenn & Segrott, 1972), have been structurally characterized. We have recently reported the palladium compounds [Pd(SCH<sub>2</sub>CH<sub>2</sub>SCH<sub>2</sub>CH<sub>2</sub>S)(PPh<sub>3</sub>)<sub>2</sub>] and [Pd<sub>2</sub>(PPh<sub>3</sub>)<sub>2</sub>(HOC<sub>6</sub>H<sub>4</sub>S)<sub>2</sub>Cl<sub>2</sub>] (Cao, Hong, Jiang, Xie & Liu, 1996), [Pd<sub>2</sub>(PPh<sub>3</sub>)<sub>2</sub>(SC<sub>2</sub>H<sub>4</sub>S)<sub>2</sub>] (Cao, Hong, Jiang & Liu, 1995) and [Pd{Ph<sub>2</sub>P(CH<sub>2</sub>)<sub>2</sub>PPh<sub>2</sub>}-(SC<sub>3</sub>H<sub>6</sub>S)].CH<sub>3</sub>CN (Su, Hong, Zhou, Xue, Liu & Mak, 1996). We report here the crystal structure

of a mononuclear complex, [Pd{Ph<sub>2</sub>P(CH<sub>2</sub>)<sub>2</sub>PPh<sub>2</sub>}-(SCH<sub>2</sub>Ph)<sub>2</sub>], (I).



(I)

The title compound is a mononuclear palladium(II) complex, where the Pd atom is four-coordinated by two phosphine P atoms and two S atoms from two PhCH<sub>2</sub>S<sup>−</sup> ligands, and has a distorted square-planar geometry. The molecule possesses a crystallographic twofold axis passing through the Pd atoms and the midpoint of C(1)—C(1<sup>i</sup>) [symmetry code: (i) 1 − *x*, −*y*, *z*]. The Pd(1), P(1), P(1<sup>i</sup>), S(1) and S(1<sup>i</sup>) atoms are in a plane with displacements of 0.0000, −0.087, 0.085, 0.073 and −0.070 Å, respectively. The Pd—S and Pd—P distances are 2.360 (2) and 2.277 (2) Å, respectively. The structure of the title compound is depicted in Fig. 1.

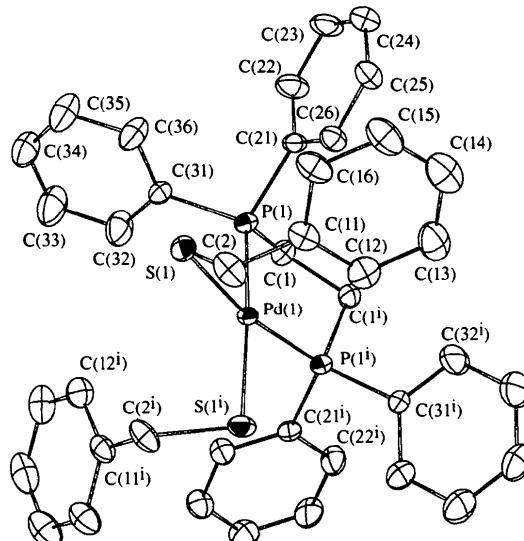


Fig. 1. The structure of [Pd{Ph<sub>2</sub>P(CH<sub>2</sub>)<sub>2</sub>PPh<sub>2</sub>}](SCH<sub>2</sub>S)<sub>2</sub>] with displacement ellipsoids at the 30% probability level.

#### Experimental

The title compound was obtained from the reaction of PdCl<sub>2</sub>, NaSCH<sub>2</sub>Ph and Ph<sub>2</sub>P(CH<sub>2</sub>)<sub>2</sub>PPh<sub>2</sub> (molar ratio 1:2:1) in MeOH, and recrystallized from CH<sub>3</sub>CN solution.

#### Crystal data

[Pd(C<sub>7</sub>H<sub>7</sub>S)<sub>2</sub>(C<sub>26</sub>H<sub>24</sub>P<sub>2</sub>)] Mo K $\alpha$  radiation  
 $M_r = 751.23$   $\lambda = 0.71073 \text{ \AA}$

Orthorhombic  
*Aba*2  
 $a = 18.070(3)$  Å  
 $b = 21.230(4)$  Å  
 $c = 9.021(2)$  Å  
 $V = 3460.9(12)$  Å<sup>3</sup>  
 $Z = 4$   
 $D_x = 1.442$  Mg m<sup>-3</sup>  
 $D_m$  not measured

*Data collection*

Enraf–Nonius CAD-4  
diffractometer  
 $\omega$ – $\theta$  scans  
Absorption correction:  
empirical via  $\psi$  scan  
(North, Phillips &  
Mathews, 1968)  
 $T_{\min} = 0.670$ ,  $T_{\max} =$   
0.923  
3370 measured reflections  
3334 independent reflections

*Refinement*

Refinement on  $F$   
 $R = 0.039$   
 $wR = 0.054$   
 $S = 1.23$   
2992 reflections  
203 parameters  
H-atom parameters not  
refined  
 $w = 1/[\sigma^2(F_o) + k(F_o^2)]$

Cell parameters from 20  
reflections  
 $\theta = 9$ – $12^\circ$   
 $\mu = 0.763$  mm<sup>-1</sup>  
 $T = 296(1)$  K  
Rectangular  
 $0.35 \times 0.20 \times 0.15$  mm  
Red

2992 observed reflections  
 $[I > 3\sigma(I)]$   
 $R_{\text{int}} = 0.032$   
 $\theta_{\text{max}} = 25^\circ$   
 $h = -21 \rightarrow 0$   
 $k = 0 \rightarrow 24$   
 $l = -10 \rightarrow 10$   
3 standard reflections  
monitored every 250  
reflections  
intensity decay: none

$(\Delta/\sigma)_{\max} = 0.01$   
 $\Delta\rho_{\max} = 0.49$  e Å<sup>-3</sup>  
 $\Delta\rho_{\min} = -0.43$  e Å<sup>-3</sup>  
Extinction correction: none  
Atomic scattering factors  
from *International Tables*  
for X-ray Crystallography  
(1974, Vol. IV)

S(1)—C(2)	1.860(7)	C(2)—C(11)	1.48(1)
P(1)—C(1)	1.834(7)	C(1)—C(1')	1.601(9)
S(1)—Pd(1)—S(1')	96.41(6)	C(21)—P(1)—C(31)	104.7(3)
S(1)—Pd(1)—P(1)	89.57(6)	S(1)—C(2)—C(11)	113.5(5)
S(1)—Pd(1)—P(1')	170.27(5)	P(1)—C(1)—C(1)	106.1(4)
P(1)—Pd(1)—P(1)	85.58(6)	C(2)—C(11)—C(12)	121.2(6)
Pd(1)—S(1)—C(2)	102.7(2)	C(2)—C(11)—C(16)	120.1(6)
Pd(1)—P(1)—C(1)	107.6(2)	P(1)—C(21)—C(22)	119.8(5)
Pd(1)—P(1)—C(21)	119.4(2)	P(1)—C(21)—C(26)	119.4(5)
Pd(1)—P(1)—C(31)	113.3(2)	P(1)—C(31)—C(32)	121.4(5)
C(1)—P(1)—C(21)	104.1(3)	P(1)—C(31)—C(36)	118.1(5)
C(1)—P(1)—C(31)	106.7(3)		

Symmetry code: (i)  $1 - x, -y, z$ .

The structure was solved by direct methods. All non-H atoms were refined by full-matrix least-squares methods (*LSFM*; B. A. Frenz & Associates, Inc., 1985), with anisotropic displacement parameters, and the calculations included H atoms. All calculations were performed on a 486PC computer with the *MolEN* (PC version; Fair, 1990) program package.

Data collection: *CAD-4 Software* (Enraf–Nonius, 1989). Cell refinement: *CAD-4 Software*. Data reduction: *CAD-4 Software*. Program(s) used to solve structure: *MULTAN82* (Main *et al.*, 1982) in *MolEN*. Molecular graphics: *ORTEPII* (Johnson, 1976). Software used to prepare material for publication: *WP5.1* and *GCIF* (local software).

This work was supported by the National Natural Scientific Foundation of China and Natural Scientific Foundation of Fujian Province.

Lists of structure factors, anisotropic displacement parameters, H-atom coordinates and complete geometry have been deposited with the IUCr (Reference: KH1116). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters (Å<sup>2</sup>)

	$x$	$y$	$z$	$B_{\text{eq}}$
Pd(1)	1/2	0	0.5312	2.074(8)
S(1)	0.53051(9)	0.07872(7)	0.3568(2)	3.19(3)
P(1)	0.54228(8)	0.06336(6)	0.7164(2)	2.44(2)
C(2)	0.4484(4)	0.0792(3)	0.2325(7)	4.0(1)
C(1)	0.5390(3)	0.0179(3)	0.8892(7)	3.0(1)
C(11)	0.3811(4)	0.1055(3)	0.3030(7)	3.4(1)
C(12)	0.3342(4)	0.0675(3)	0.3898(9)	4.3(2)
C(13)	0.2708(4)	0.0921(4)	0.458(1)	5.4(2)
C(14)	0.2529(5)	0.1558(4)	0.433(1)	5.8(2)
C(15)	0.2983(5)	0.1929(4)	0.346(1)	5.6(2)
C(16)	0.3625(4)	0.1686(3)	0.2824(9)	4.5(2)
C(21)	0.4924(3)	0.1361(3)	0.7603(8)	2.7(1)
C(22)	0.5189(4)	0.1761(3)	0.873(1)	4.5(1)
C(23)	0.4818(4)	0.2334(3)	0.899(1)	4.7(2)
C(24)	0.4201(4)	0.2489(3)	0.8198(9)	4.2(2)
C(25)	0.3935(4)	0.2085(3)	0.7107(9)	3.8(1)
C(26)	0.4296(4)	0.1512(3)	0.6817(8)	3.2(1)
C(31)	0.6375(3)	0.0878(3)	0.6913(7)	2.7(1)
C(32)	0.6955(4)	0.0476(4)	0.721(1)	5.4(2)
C(33)	0.7696(4)	0.0665(4)	0.695(1)	7.0(2)
C(34)	0.7820(4)	0.1249(4)	0.632(1)	5.8(2)
C(35)	0.7240(4)	0.1659(4)	0.596(1)	6.1(2)
C(36)	0.6510(4)	0.1475(4)	0.628(1)	4.9(2)

Table 2. Selected geometric parameters (Å, °)

Pd(1)—S(1)	2.360(2)	P(1)—C(21)	1.832(6)
Pd(1)—P(1)	2.277(2)	P(1)—C(31)	1.810(6)

**References**

- B. A. Frenz & Associates, Inc. (1985). *LSFM*. College Station, Texas, USA, and Enraf–Nonius, Delft, The Netherlands.
- Cao, R., Hong, M. C., Jiang, F. L. & Liu, H. Q. (1995). *Acta Cryst. C51*, 1280–1282.
- Cao, R., Hong, M. C., Jiang, F. L., Xie, X. L. & Liu, H. Q. (1996). *Polyhedron*, **15**, 4047–4051.
- Enraf–Nonius (1989). *CAD-4 Software*. Version 5.0. Enraf–Nonius, Delft, The Netherlands.
- Fair, C. K. (1990). *MolEN. An Interactive Intelligent System for Crystal Structure Analysis*. Enraf–Nonius, Delft, The Netherlands.
- Fenn, R. H. & Segrott, G. R. (1972). *J. Chem. Soc. Dalton Trans.*, pp. 330–333.
- Johnson, C. K. (1976). *ORTEPII*. Report ORNL-5138. Oak Ridge National Laboratory, Tennessee, USA.
- Main, P., Fiske, S. J., Hull, S. E., Lessinger, L., Germain, G., Declercq, J.-P. & Woolfson, M. M. (1982). *MULTAN11/82. A System of Computer Programs for the Automatic Solution of Crystal Structures from X-ray Diffraction Data*. Universities of York, England, and Louvain, Belgium.
- North, A. C. T., Phillips, D. C. & Mathews, F. S. (1968). *Acta Cryst. A24*, 351–359.
- Su, W.-P., Hong, M.-C., Zhou, Z.-Y., Xue, F., Liu, H.-Q. & Mak, T. C. W. (1996). *Acta Cryst. C52*, 2691–2693.